

INDOLE DERIVATIVES

LXXXII.* SYNTHESIS OF N¹⁵-INDOLE DERIVATIVES

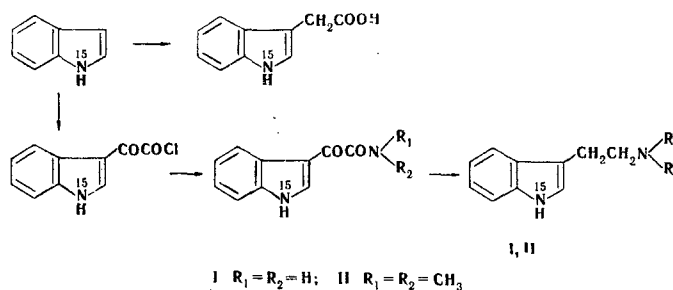
L. I. Dmitrevskaya, Yu. I. Smushkevich,
A. D. Pozdnyakov, and N. N. Suvorov

UDC 547.753.757.07+546.79

The synthesis of N¹⁵-3-indolylacetic acid, 1-N¹⁵-tryptamine, and 1-N¹⁵-N,N-dimethyltryptamine was accomplished.

The synthesis of N¹⁵-indole by cyclization of 1-N¹⁵-phenylhydrazone under heterogeneous catalysis conditions on γ -aluminum oxide [2] opens up a convenient route to the preparation of biologically active indole derivatives with a labeled nitrogen.

In the present study we have accomplished the synthesis of N¹⁵-3-indolylacetic acid, 1-N¹⁵-tryptamine, and 1-N¹⁵-N,N-dimethyltryptamine.



N¹⁵-3-Indolylacetic acid was obtained by heating N¹⁵-indole with chloroacetic acid at 245–250°C in the presence of KOH in an autoclave, while 1-N¹⁵-tryptamine (I) and 1-N¹⁵-N,N-dimethyltryptamine (II) were obtained by reduction of the corresponding 1-N¹⁵-3-indolylloxalylamides with lithium aluminum hydride.

The substances were analyzed for their N¹⁵ content by combustion to nitrogen by the Dumas method [3] with subsequent mass-spectrometric determination of the percentage of N¹⁵ (Table 1).

EXPERIMENTAL

The isotopic analysis for the percentage of N¹⁵ was performed with an MI-1305 mass spectrometer.

N¹⁵-3-Indolylacetic Acid. A 100-ml autoclave was charged with 1.2 g (0.01 mole) of N¹⁵-indole, 1.44 g (0.015 mole) of chloroacetic acid, 6.15 g of KOH, and 30 ml of water. The autoclave was then evacuated and filled with argon to 5 atm. The mixture was then heated at 245–250° with shaking for 12 h, after which it was filtered, and the filtrate was cooled to 6° and acidified (with respect to Congo Red) with 16% HCl. The resulting precipitate was removed by filtration and recrystallized from water to give 1.14 g (64%) of a product with mp 164–165° (mp 164–165° [4]).†

*See [1] for communication LXXXI.

†Here and elsewhere, the physical constants of the corresponding compounds with a natural isotopic composition are presented for comparison.

D. I. Mendeleev Chemical-Engineering Institute, Moscow. Translated from Zhurnal Geterotsiklicheskih Soedinenii, No. 4, pp. 516–517, April, 1973. Original article submitted April 14, 1972.

© 1975 Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.

TABLE 1. Results of the Determination of the Percent N¹⁵

Substance	N ¹⁵ , %
N ¹⁵ -Indole	10.9
N ¹⁵ -3-Indolylacetic acid	10.7
1-N ¹⁵ -Tryptamine	5.33
1-N ¹⁵ -N,N-Dimethyltryptamine	5.46

N¹⁵-3-Indolyloxalyl Chloride. A 1.2-g (0.01 mole) sample of N¹⁵-indole was dissolved in 25 ml of absolute ether, the solution was cooled to -5°, and 1.3 g (0.01 mole) of oxalyl chloride was added slowly by drops. The mixture was then stirred for 30 min, and the precipitate was removed by filtration and washed with ether to give 1.8 g (90%) of a product with mp 136° (dec.) (mp 135-136° (dec.) [5]).

1-N¹⁵-3-Indolyloxalylamide. A 1.8-g (8.7 mmole) sample of 3-indolyloxalyl chloride was added to 30 ml of a saturated aqueous solution of ammonia, and the mixture was stirred at 60-70° for 1.5 h. The precipitate was removed by filtration to give 1.55 g (94%) of a product with mp 251° (dec., from alcohol) (mp 251-252° (dec.) [5]).

1-N¹⁵-N,N-Dimethyl-3-indolyloxalylamide. This compound was similarly obtained in 90% yield and had mp 158-159° (from benzene) (mp 160-162° [6]).

1-N¹⁵-Tryptamine. A solution of 0.7 g (3.7 mmole) of 1-N¹⁵-3-indolyloxalylamide in 20 ml of absolute tetrahydrofuran was added by drops to 1 g (0.026 mole) of LiAlH₄ in 30 ml of absolute tetrahydrofuran, after which the mixture was stirred and refluxed for 4 h. It was then cooled, and 1 ml of water, 1 ml of 15% NaOH, and 3 ml of water were added successively. The mixture was then stirred for 3 h, and the precipitate was removed by filtration and washed with tetrahydrofuran. The filtrate was dried with MgSO₄, the solvent was removed by vacuum distillation, and the residue was dissolved in methylene chloride and acidified with 5% HCl in alcohol. The precipitated hydrochloride of I was removed by filtration and washed on the filter with ether to give 0.4 g (55%) of a product with mp 240-242° (mp 243-244° [7]).

1-N¹⁵-N,N-Dimethyltryptamine. This compound was similarly obtained except that the amine was isolated as the oxalate. The yield of product with mp 148-149° (mp 151-152° [6]) was 60%.

LITERATURE CITED

1. I. D. Pletnev, V. F. Alyautdinova, and N. N. Suvorov, *Khim. Geterotsikl. Soedin.*, 1632 (1972).
2. N. N. Suvorov, L. I. Dmitrevskaya, Yu. I. Smushkevich, and A. D. Pozdnyakov, *Zh. Obshch. Khim.*, **42**, 2746 (1972).
3. J. Beynon, *Mass Spectrometry and Its Application to Organic Chemistry*, American Elsevier (1960).
4. F. E. King and P. L. Ecuyer, *J. Chem. Soc.*, 1903 (1934).
5. M. S. Kharasch, S. K. Stephen, and H. C. Brown, *J. Am. Chem. Soc.*, **62**, 2242 (1940).
6. H. Kondo, H. Kataoka, I. Hayashi, and T. Dodo, *Itsuo Kenkyusho Nempo*, **10**, 1 (1959); *Chem. Abstr.*, **54**, 492 (1960).
7. M. Ondo, M. Kawanishi, and M. Sasamoto, *J. Pharm. Soc. Japan*, **76**, 409 (1956).